

4-Functionally-Substituted 3-Heterylpyrazoles: XVII.* 3-Aryl-1-phenyl-4-pyrazolmethylsulfanyl(sulfinyl, sulfonyl)acetic Acids

M.K. Bratenko^a, V.A. Chornous^a, and M.V. Vovk^b

^aBukovinskaya State Medical Academy, Chernovtsy, Ukraine

^bInstitute of Organic Chemistry, National Academy of Sciences of Ukraine, Kiev, 02094, Ukraine

e-mail: mvovk@i.com.ua

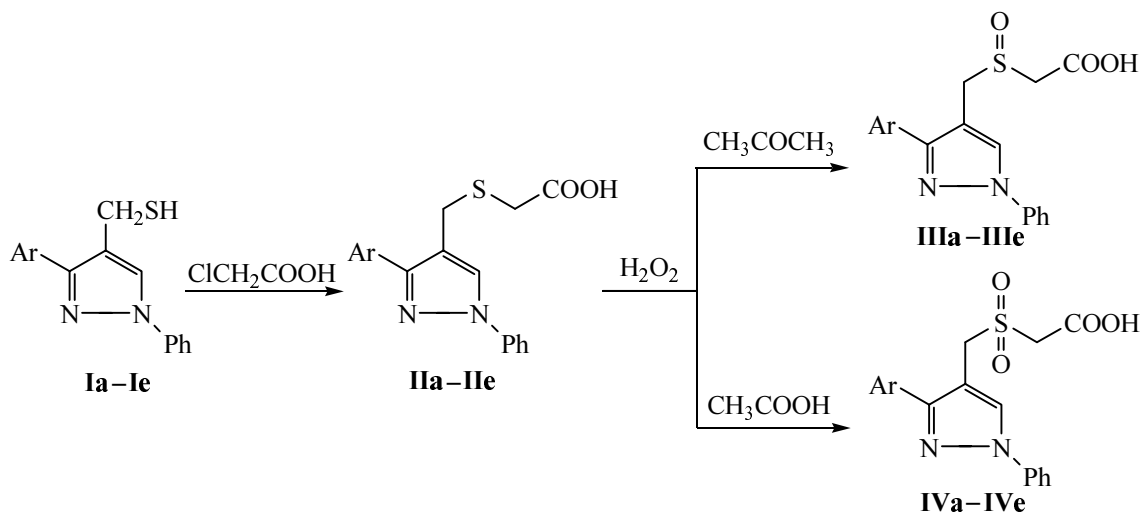
Received March 5, 2005

Abstract—3-Aryl-1-phenyl-4-mercaptomethylpyrazoles reacted with monochloroacetic acid to give 3-aryl-1-phenyl-4-pyrazolmethylsulfanylacetic acids whose oxidation with hydrogen peroxide in acetone or acetic acid solution led to 3-aryl-1-phenyl-4-pyrazolmethylsulfinyl- and sulfonylacetic acids, respectively.

DOI: 10.1134/S1070428006050101

Sulfur-containing acetic acids possessing versatile synthetic possibilities and useful properties are a matter of rapt attention of chemists. By now a thorough investigation was carried out on transformations of a series of benzylsulfanylacetic acids into their sulfinyl and sulfonyl derivatives [2–8]. The benzylsulfanylacetic acids proper turned out to be convenient models for studying the optical properties of sulfoxide systems [6–8], and benzylsulfonylacetic acids found the use as important blocks in the synthesis of α,β -unsaturated sulfones [9, 10].

We report here on the results of the study on the synthesis of previously unknown heterocyclic analogs of the above mentioned sulfur-containing compounds, 4-pyrazolmethylthioacetic acids and their oxidation products. We chose as initial compounds 4-mercaptomethylpyrazoles **Ia–If** that we had synthesized previously [11]. These substances in the presence of alcoholic alkali were readily alkylated with the monochloroacetic acid to afford 4-pyrazolmethylsulfanylacetic acids **IIa–IIf** in 62–73% yields. The latter were identified based on IR,



Ar = Ph (**a**), 4- FC_6H_4 (**b**), 4- ClC_6H_4 (**c**), 4- BrC_6H_4 (**d**), 4- MeOC_6H_4 (**e**), 3,4- $\text{Me}_2\text{C}_6\text{H}_3$ (**f**).

* For Communication XVI, see [1].

^1H NMR spectra, and elemental analysis. In particular, the ^1H NMR spectra contained characteristic singlets belonging to the methylene protons of the carboxymethyl (3.17–3.26) and pyrazolemethyl (3.88–3.92 ppm) fragments.

We found conditions for a selective conversion of acids **IIa–IIc** into 4-pyrazolymethylsulfanylacetic acids **IIIa–IIIc** and 4-pyrazolymethyl-sulfonyl-acetic acids **IVa–IVc**. It was shown that the treatment of acetone solutions of compounds **IIa–IIc** with 30% hydrogen peroxide at room temperature furnished in 78–92% yields acids **IIIa–IIIc**. In their turn, acids **IVa–IVc** were obtained in 85–98% yields by oxidation of compounds **IIa–IIc** with 30% hydrogen peroxide in acetic acid at 60°C.

The characteristic feature of acids **III** is the nonequivalence of the protons in both methylene groups caused by the pyramidal structure of the sulfoxide moiety. In the ^1H NMR spectra these protons give rise to quartets of *AB* systems in the range 3.63–3.98 and 4.15–4.36 ppm respectively. At the same time the methylene groups of acids **IVa–IVc** appear in the ^1H NMR spectra as two singlets: CH_2COOH at 4.30–4.35, PyrCH_2 at 4.66–4.70 ppm. In general the oxidation of sulfides **II** into sulfoxides **III** and sulfones **IV** tends to increase the acid character of the methylene groups that is revealed in the ^1H NMR spectra by their successive downfield shift by approximately 0.4–0.6 ppm.

EXPERIMENTAL

IR spectra of compounds were recorded on a spectrophotometer UR-20 from KBr pellets. ^1H NMR spectra were registered from solutions of compounds in $(\text{CD}_3)_2\text{SO}$ on a spectrometer Varian Gemini (300 MHz), internal reference TMS.

3-Aryl-1-phenyl-4-pyrazolymethylsulfanylacetic acid IIa–IIc. A mixture of 10 mmol of 4-mercapto-methylpyrazole **Ia–Ic**, 1 g (10.6 mmol) of monochloroacetic acid, and 10 ml of 20% water solution of NaOH in 10 ml of ethanol was boiled for 2 h, cooled, then 50 ml of water was added, and the solution was filtered. The filtrate was acidified with 6 N hydrochloric acid, the arising oily substance get solid within 24 h and was recrystallized from 80% aqueous ethanol.

1,3-Diphenyl-4-pyrazolymethylsulfanylacetic acid (IIa). Yield 63%, mp 87–88°C. IR spectrum, cm^{-1} : 1720 (C=O), 2650–2940 (OH). ^1H NMR spectrum, δ , ppm: 3.23 s (2H, CH_2COOH), 3.91 s (2H, PyrCH_2), 7.26–7.85 m (10 H_{arom}), 8.44 s (1H, H_{Pyr}^5), 12.51 br.s (1H,

COOH). Found, %: C 66.42; H 4.81; N 8.56. $\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}_2\text{S}$. Calculated, %: C 66.67; H 4.94; N 8.64.

1-Phenyl-3-(4-fluorophenyl)-4-pyrazolymethyl-sulfanylacetic acid (IIb). Yield 71%, mp 108–111°C. IR spectrum, cm^{-1} : 1725 (C=O), 2680–2950 (OH). ^1H NMR spectrum, δ , ppm: 3.26 s (2H, CH_2COOH), 3.91 s (2H, PyrCH_2), 7.25–7.87 m (9 H_{arom}), 8.45 s (1H, H_{Pyr}^5), 12.47 br.s (1H, COOH). Found, %: C 63.40; H 4.51; N 7.98. $\text{C}_{18}\text{H}_{15}\text{FN}_2\text{O}_2\text{S}$. Calculated, %: C 63.16; H 4.39; N 8.19.

1-Phenyl-3-(4-chlorophenyl)-4-pyrazolymethyl-sulfanylacetic acid (IIc). Yield 68%, mp 119–120°C. IR spectrum, cm^{-1} : 1720 (C=O), 2650–2930 (OH). ^1H NMR spectrum, δ , ppm: 3.17 s (2H, CH_2COOH), 3.88 s (2H, PyrCH_2), 7.27–7.88 m (9 H_{arom}), 8.45 s (1H, H_{Pyr}^5), 12.40 br.s (1H, COOH). Found, %: C 60.03; H 3.99; N 7.67. $\text{C}_{18}\text{H}_{15}\text{ClN}_2\text{O}_2\text{S}$. Calculated, %: C 60.25; H 4.18; N 7.81.

3-(4-Bromophenyl)-1-phenyl-4-pyrazolymethylsulfanylacetic acid (IIc). Yield 73%, mp 138–140°C. IR spectrum, cm^{-1} : 1725 (C=O), 2700–2980 (OH). ^1H NMR spectrum, δ , ppm: 3.26 s (2H, CH_2COOH), 3.92 s (2H, PyrCH_2), 7.30–7.90 m (9 H_{arom}), 8.46 s (1H, H_{Pyr}^5), 12.52 br.s (1H, COOH). Found, %: C 53.33; H 3.60; N 6.81. $\text{C}_{18}\text{H}_{15}\text{BrN}_2\text{O}_2\text{S}$. Calculated, %: C 53.60; H 3.72; N 6.95.

3-(4-Methoxy)-1-phenyl-4-pyrazolymethyl-sulfanylacetic acid (IIe). Yield 65%, mp 110–111°C. IR spectrum, cm^{-1} : 1720 (C=O), 2670–2920 (OH). ^1H NMR spectrum, δ , ppm: 3.26 s (2H, CH_2COOH), 3.81 s (3H, CH_3O), 3.90 s (2H, PyrCH_2), 6.99 d (2 H_{arom}), 7.26 t (1 H_{arom}), 7.46 t (2 H_{arom}), 7.74 d (2 H_{arom}), 7.86 d (2 H_{arom}), 8.39 s (1H, H_{Pyr}^5), 12.57 br.s (1H, COOH). Found, %: C 64.16; H 4.91; N 7.78. $\text{C}_{19}\text{H}_{18}\text{N}_2\text{O}_3\text{S}$. Calculated, %: C 64.41; H 5.08; N 7.91.

3-(3,4-Dimethylphenyl)-1-phenyl-4-pyrazolymethylsulfanylacetic acid (IIc). Yield 62%, mp 99–101°C. IR spectrum, cm^{-1} : 1725 (C=O), 2650–2950 (OH). ^1H NMR spectrum, δ , ppm: 2.28 s (3H, CH_3), 2.31 s (3H, CH_3), 3.23 s (2H, CH_2COOH), 3.88 s (2H, PyrCH_2), 7.21–7.59 m (6 H_{arom}), 7.86 d (2 H_{arom}), 8.40 s (1H, H_{Pyr}^5), 12.40 br.s (1H, COOH). Found, %: C 68.44; H 5.56; N 7.81. $\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$. Calculated, %: C 68.18; H 5.68; N 7.95.

3-Aryl-1-phenyl-4-pyrazolymethylsulfanylacetic acids IIIa–IIIc. To a solution of 5 mmol of acid **IIa–IIc** in 20 ml of acetone was added 3 ml of 30% solution of

hydrogen peroxide, and the mixture was stirred for 3 days at room temperature. The solution was evaporated, the residue was recrystallized from 75% acetic acid.

1,3-Diphenyl-4-pyrazolylmethylsulfanylacetic acid (IIIa). Yield 83%, mp 138–140°C. IR spectrum, cm^{-1} : 1730 (C=O), 2750–2920 (OH). ^1H NMR spectrum, δ , ppm: 3.64 d (1H, CH_2COOH , 2J 13 Hz), 3.97 d (1H, CH_2COOH , 2J 13 Hz), 4.17 d (1H, PyrCH_2 , 2J 14 Hz), 4.30 d (1H, PyrCH_2 , 2J 14 Hz), 7.28–7.89 m (10 H_{arom}), 8.57 s (1H, H_{Pyr}^5), 13.03 br.s (1H, COOH). Found, %: C 63.80; H 4.68; N 8.42. $\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}_3\text{S}$. Calculated, %: C 63.53; H 4.71; N 8.23.

1-Phenyl-3-(4-fluorophenyl)-4-pyrazolylmethylsulfanylacetic acid (IIIb). Yield 90%, mp 183–185°C. IR spectrum, cm^{-1} : 1735 (C=O), 2720–2950 (OH). ^1H NMR spectrum, δ , ppm: 3.63 d (1H, CH_2COOH , 2J 14 Hz), 3.93 d (1H, CH_2COOH , 2J 14 Hz), 4.16 d (1H, PyrCH_2 , 2J 14 Hz), 4.28 d (1H, PyrCH_2 , 2J 14 Hz), 7.24–7.87 m (9 H_{arom}), 8.53 s (1H, H_{Pyr}^5), 12.87 br.s (1H, COOH). Found, %: C 60.07; H 4.32; N 8.03. $\text{C}_{18}\text{H}_{15}\text{FN}_2\text{O}_3\text{S}$. Calculated, %: C 60.34; H 4.19; N 7.82.

1-Phenyl-3-(4-chlorophenyl)-4-pyrazolylmethylsulfanylacetic acid (IIIc). Yield 92%, mp 178–180°C. IR spectrum, cm^{-1} : 1735 (C=O), 2700–2930 (OH). ^1H NMR spectrum, δ , ppm: 3.68 d (1H, CH_2COOH , 2J 13 Hz), 3.95 d (1H, CH_2COOH , 2J 13 Hz), 4.18 d (1H, PyrCH_2 , 2J 14 Hz), 4.33 d (1H, PyrCH_2 , 2J 14 Hz), 7.32–7.53 m (5 H_{arom}), 7.79 d (2 H_{arom}), 7.88 d (2 H_{arom}), 8.56 s (1H, H_{Pyr}^5), 13.09 br.s (1H, COOH). Found, %: C 57.94; H 4.17; N 7.68. $\text{C}_{18}\text{H}_{15}\text{ClN}_2\text{O}_3\text{S}$. Calculated, %: C 57.68; H 4.01; N 7.48.

3-(4-Bromophenyl)-1-phenyl-4-pyrazolylmethylsulfanylacetic acid (IIId). Yield 89%, mp 189–191°C. IR spectrum, cm^{-1} : 1730 (C=O), 2700–2900 (OH). ^1H NMR spectrum, δ , ppm: 3.64 d (1H, CH_2COOH , 2J 12 Hz), 3.94 d (1H, CH_2COOH , 2J 12 Hz), 4.15 d (1H, PyrCH_2 , 2J 13 Hz), 4.31 d (1H, PyrCH_2 , 2J 13 Hz), 7.32–7.93 m (9 H_{arom}), 8.55 s (1H, H_{Pyr}^5), 13.01 br.s (1H, COOH). Found, %: C 51.79; H 3.51; N 6.60. $\text{C}_{18}\text{H}_{15}\text{BrN}_2\text{O}_3\text{S}$. Calculated, %: C 51.55; H 3.58; N 6.68.

3-(4-Methoxyphenyl)-1-phenyl-4-pyrazolylmethylsulfanylacetic acid (IIIe). Yield 81%, mp 156–157°C. IR spectrum, cm^{-1} : 1725 (C=O), 2680–2900 (OH). ^1H NMR spectrum, δ , ppm: 3.65 d (1H, CH_2COOH , 2J 14 Hz), 3.82 s (3H, CH_3O), 3.98 d (1H, CH_2COOH , 2J 14 Hz), 4.19 d (1H, PyrCH_2 , 2J 15 Hz), 4.33 d (1H, PyrCH_2 , 2J 15 Hz), 7.03 d (3 H_{arom}), 7.30 t (1 H_{arom}), 7.51 t (2 H_{arom}), 7.65 d (2 H_{arom}), 7.83 d (2 H_{arom}), 8.52 s

(1H, H_{Pyr}^5), 12.93 br.s (1H, COOH). Found, %: C 61.60; H 3.59; N 7.50. $\text{C}_{19}\text{H}_{18}\text{N}_2\text{O}_4\text{S}$. Calculated, %: C 61.61; H 4.90; N 7.56.

3-(3,4-Dimethylphenyl)-1-phenyl-4-pyrazolylmethylsulfanylacetic acid (III f). Yield 78%, mp 141–143°C. IR spectrum, cm^{-1} : 1725 (C=O), 2650–2890 (OH). ^1H NMR spectrum, δ , ppm: 2.28 C (3H, CH_3), 2.31 C (3H, CH_3), 3.64 d (1H, CH_2COOH , 2J 14 Hz), 3.96 d (1H, 2J 14 Hz), 4.20 d (1H, PyrCH_2 , 2J 14 Hz), 4.36 d (1H, PyrCH_2 , 2J 14 Hz), 7.28–7.64 d (6 H_{arom}), 7.87 d (2 H_{arom}), 8.59 C (1H, H_{Pyr}^5), 12.94 br.s (1H, COOH). Found, %: C 65.35; H 5.40; N 7.50. $\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_3\text{S}$. Calculated, %: C 65.20; H 5.47; N 7.60.

3-Aryl-1-phenyl-4-pyrazolylmethylsulfanylacetic acids IVa–IVf. To a solution of 5 mmol of acid IIa–II f in 20 ml of acetic acid was added 6 ml of 30% solution of hydrogen peroxide, and the mixture was stirred for 2 h at room temperature, then 2 h more at 60°C. The reaction mixture was cooled, diluted with 50 ml of water, the separated precipitate was filtered off and recrystallized from 20% acetic acid.

1,3-Diphenyl-4-pyrazolylmethylsulfonylacetic acid (IVa). Yield 93%, mp 156–157°C. IR spectrum, cm^{-1} : 1730 (C=O), 2560–2850 (OH). ^1H NMR spectrum, δ , ppm: 4.31 s (2H, CH_2COOH), 4.69 d (2H, PyrCH_2), 7.29–7.83 m (10 H_{arom}), 8.53 s (1H, H_{Pyr}^5), 13.03 br.s (1H, COOH). Found, %: C 60.91; H 4.63; N 7.99. $\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}_4\text{S}$. Calculated, %: C 60.67; H 4.49; N 7.86.

1-Phenyl-3-(4-fluorophenyl)-4-pyrazolylmethylsulfonylacetic acid (IVb). Yield 97%, mp 178–179°C. IR spectrum, cm^{-1} : 1730 (C=O), 2590–2820 (OH). ^1H NMR spectrum, δ , ppm: 4.33 s (2H, CH_2COOH), 4.68 s (2H, PyrCH_2), 7.26–7.89 m (9 H_{arom}), 8.55 s (1H), 13.03 br.s (1H, COOH). Found, %: C 57.43; H 4.20; N 7.26. $\text{C}_{18}\text{H}_{15}\text{FN}_2\text{O}_4\text{S}$. Calculated, %: C 57.75; H 4.01; N 7.49.

1-Phenyl-3-(4-chlorophenyl)-4-pyrazolylmethylsulfonylacetic acid (IVc). Yield 95%, mp 183–185°C. IR spectrum, cm^{-1} : 1725 (C=O), 2550–2860 (OH). ^1H NMR spectrum, δ , ppm: 4.34 s (2H, CH_2COOH), 4.69 s (2H, PyrCH_2), 7.34–7.69 m (5 H_{arom}), 7.69 d (2 H_{arom}), 7.86 d (2 H_{arom}), 8.59 s (1H, H_{Pyr}^5), 12.99 br.s (1H, COOH). Found, %: C 55.06; H 3.97; N 7.34. $\text{C}_{18}\text{H}_{15}\text{ClN}_2\text{O}_4\text{S}$. Calculated, %: C 55.31; H 3.84; N 7.17.

3-(4-Bromophenyl)-1-phenyl-4-pyrazolylmethylsulfonylacetic acid (IVd). Yield 98%, mp 195–197°C. IR spectrum, cm^{-1} : 1730 (C=O), 2580–2900 (OH). ^1H NMR spectrum, δ , ppm: 4.35 s (2H, CH_2COOH),

4.70 s (2H, PyrCH₂), 7.33–7.64 m (5H_{arom}), 7.74 d (2H_{arom}), 7.89 d (2H_{arom}), 8.57 s (1H, H⁵_{Pyr}), 13.07 br.s (1H, COOH). Found, %: C 49.95; H 3.28; N 6.61. C₁₈H₁₅BrN₂O₄S. Calculated, %: C 49.66; H 3.45; N 6.44.

3-(4-Methoxyphenyl)-1-phenyl-4-pyrazolyl-methylsulfonylacetic acid (IVe). Yield 89%, mp 164–165°C. IR spectrum, cm⁻¹: 1730 (C=O), 2580–2910 (OH). ¹H NMR spectrum, δ, ppm: 3.81 s (3H, CH₃O), 4.30 s (2H, CH₂COOH), 4.68 s (2H, PyrCH₂), 6.99 d (2H_{arom}), 7.31 t (1H_{arom}), 7.50 t (2H_{arom}), 7.70 d (2H_{arom}), 7.90 d (2H_{arom}), 8.57 s (1H, H⁵_{Pyr}), 13.07 br.s (1H, COOH). Found, %: C 49.95; H 3.28; N 6.61. C₁₈H₁₅BrN₂O₄S. Calculated, %: C 49.66; H 3.45; N 6.44.

3-(3,4-Dimethylphenyl)-1-phenyl-4-pyrazolyl-methylsulfonylacetic acid (IVf). Yield 85%, mp 148–150°C. IR spectrum, cm⁻¹: 1730 (C=O), 2610–2890 (OH). ¹H NMR spectrum, δ, ppm: 2.25 s (3H, CH₃), 2.28 s (3H, CH₃), 4.32 s (2H, CH₂COOH), 4.67 s (2H, PyrCH₂), 7.21–7.59 m (6H_{arom}), 7.90 d (2H_{arom}), 8.52 s (1H, H⁵_{Pyr}), 12.93 br.s (1H, COOH). Found, %: C 62.75; H 5.04; N 7.47. C₂₀H₂₀N₂O₄S. Calculated, %: C 62.50; H 5.21; N 7.29.

REFERENCES

1. Bratenko, M.K., Chornous, V.A., and Vovk, M.V., *Zh. Org. Khim.*, 2006, vol. 42, p. 701.
2. Lesser, R. and Mehrlander, A., *Chem. Ber.*, 1923, vol. 56, p. 1647.
3. Stoner, G.G. and Dougherty, G., *J. Am. Chem. Soc.*, 1941, vol. 63, p. 1481.
4. Aroyan, A.A. and Antonyan, R.K., *Arm. Khim. Zh.*, 1973, vol. 23, p. 369.
5. Newman, M.S., Fones, W., and Renoll, M., *J. Am. Chem. Soc.*, 1947, vol. 69, p. 718.
6. Janczewski, M. and Janowski, W., *Roczn. Chem.*, 1975, vol. 49, p. 1961.
7. Janczewski, M., Ksiezopolski, J., and Rak-Najda, T., *Pol. J. Chem.*, 1981, vol. 55, p. 535.
8. Janczewski, M. and Ksiezopolski, J., *Pol. J. Chem.*, 1984, vol. 58, p. 103.
9. Balasubramanian, M., Baliah, V., and Rangarajan, T., *J. Chem. Soc.*, 1955, p. 3296.
10. Sharma, V.M., Adi, Seshu, K.V., Madan, S., Vishnu, P., Babu, A., Krishna, V.C., Sreenu, J., Krishna, R.V., Venkateswarlu, A., Rajagopal, S., Ajaykumar, R., and Kumar, S.T., *Bioorg. Med. Chem. Lett.*, 2004, vol. 14, p. 67.
11. Bratenko, M.K., Chornous, V.A., and Vovk, M.V., *Zh. Org. Khim.*, 2002, vol. 38, p. 432.